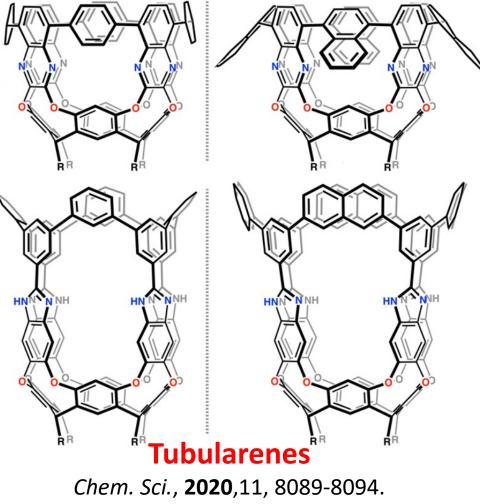


Fluorocages: C–H hydrogen bonding for the recognition of anions

Saber Mirzaei April 6, 2021

## Rational design of anion hosts to solve a problem

What we do in our lab:



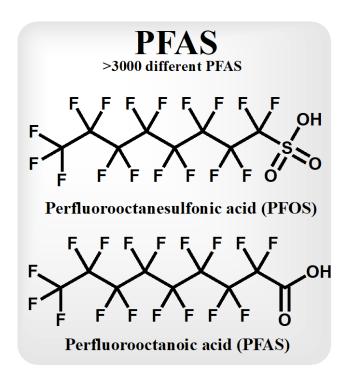
*Org. Lett.* **2021**, 23, 1, 87–92.

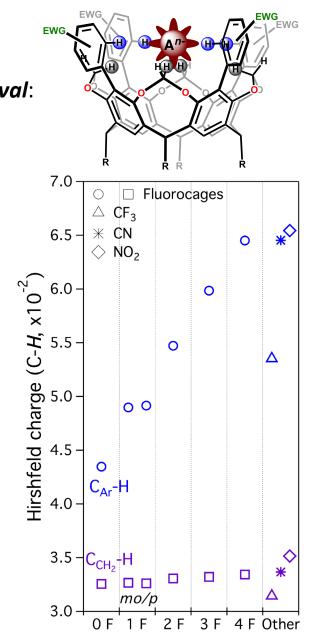
Best current technologies for PFAS removal:
1) Burn contaminated system

\*Activated carbon

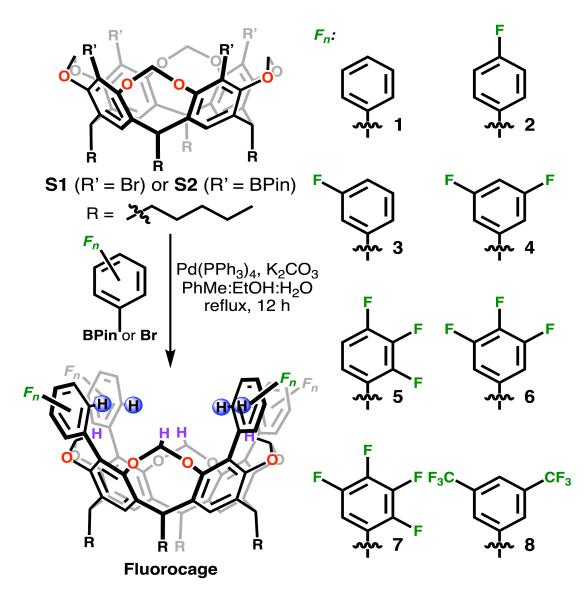
2) Bury contaminated system
3) Other traditional systems to

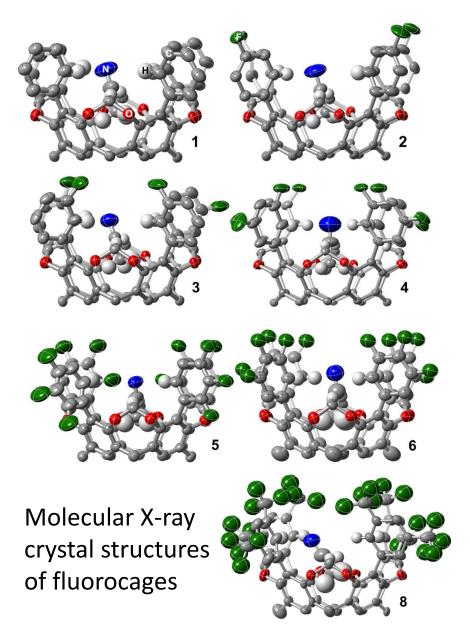
clean water not designed for PFAS





### Synthesis & Characterization





# Do they work?

Anion association constants ( $K_a$ ) of  $[PF_6]^-$  and  $[MeSO_3]^-$  in fluorocages 1–8. Ka Solvent Salt 1 2 5 6 7 8 3 4 [*n*-Bu₄N]  $15105\pm$ 0.0 0.0 0.0 84±12 ND ND 279±15 328  $[\mathsf{PF}_6]$ [*n*-Bu₄N] CDCl<sub>3</sub> TBD 0.0 22±4 667±38 TBD ND TBD ND [MeSO<sub>2</sub>]



#### ND = Not determined, TBD = To be determined

#### 7.0 Fluorocages 0 △ CF<sub>3</sub> 6.5 -× CΝ \* 0 $\Diamond NO_2$ Hirshfeld charge (C-*H*, x10<sup>-2</sup>) 0 0 00 I C<sub>Ar</sub>-F **Top View** Side View 3.5 - C<sub>CH</sub>, - F Single-crystal structure of $MeSO_3 \subset 6$ cage 3.0-0 F 1 F 2 F 3 F 4 F Other

### Boss: Raúl Hernández Sánchez



Students: Saber and Victor





Arts & Sciences Graduate Fellowship